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Isolation and characterization of cytotoxic compounds from *Euphorbia cornigera* Boiss.

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Methanolic extract of *Euphorbia cornigera* shoots was separated using HPLC, affording compounds **1**–**4**. Their structures and relative stereochemistry were established after obtaining their spectroscopic (IR, ^1H , ^{13}C NMR COSY-45°, HOHAHA, HSQC, HMBC, NOESY, and mass measurement) data. On the basis of these data, the compounds were characterized as 3-*O*-(2,3-dimethylbutanoyl)-13-*O*-dodecanoyl-20-*O*-tetradecanoylingenol (**1**), 3-*O*-decanoyl-20-*O*-hexanoylingenol (**2**), 3-*O*-(2,3-dimethylbutanoyl)-13-*O*-dodecanoyl-20-*O*-hexadecanoylingenol (**3**), and 13-*O*-dodecanoyl-20-*O*-hexanoylingenol (**4**); among these compounds, two (**1** and **2**) were new metabolites while the rest (**3** and **4**) were known. The MTT cytotoxicity assay was carried out using amrubicin hydrochloride as a positive control. Compound **1** displayed IC₅₀ as 5.0 and 2.9 μM against RAW and HT-29 cell lines, respectively, which is 5- and 1.5-folds stronger than the control with IC₅₀ values of 25 and 4.36 μM, respectively.

Keywords: *Euphorbia cornigera* Boiss.; Euphorbiaceae; cytotoxicity; ingenol

1. Introduction

Importance of plants is well recognized as the valuable source of novel anti-cancer and other drugs throughout medical history [1–5]. Most of the anti-cancer drugs such as vinblastine, vincristine, taxol, camptothecin, topotecan, irinotecan, podophyllotoxin, etoposide, and teniposide are derived from plants [1]. However, many tumors are showing resistance against available drugs [6,7] and hence these drugs have only limited anti-solid tumor activities [6,8,9]. Therefore, there is a need to discover new plant-derived anti-cancer drugs, which may have more potential than the present drugs or may show a different mechanism to cure tumors.

Fortunately, Euphorbiaceae family, which is one of the largest families of plants, comprising 300 genera and 7500

species, is commonly available in plain, mountainous, and desert areas of Pakistan [10,11]. These plants are popular among the local people for the treatment of various ailments including cancer [11–13]. The Euphorbiaceae family is also famous for its bio-diversity and enrichment in highly toxic, carcinogenic, co-carcinogenic, tumor-promoter to nontoxic, anti-tumor larvicidal, and molluscicidal compounds [12–14]. These aspects prompted us to investigate indigenous medicinal plants of Pakistan. In this paper, two new (**1** and **2**) and two known (**3** and **4**) cytotoxic compounds from *Euphorbia cornigera* are presented.

2. Results and discussion

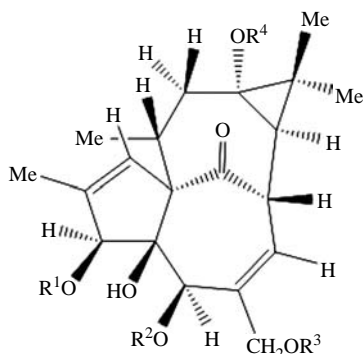
The MeOH extract of *E. cornigera* shoots after dispersing in hot water was further

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extracted with Et₂O, which showed promising cytotoxicity. The extract was subjected to silica gel column chromatography to obtain semi-pure active material. After separation on HPLC, the active portion afforded compounds **1–4** (Figure 1).

The active fraction of the ether extract (compounds **1–4**) was hydrolyzed with 0.5 M HCl and was separated through HPLC. The product was identified through GC-MS and (¹H, ¹³C) NMR spectrometry (Table 1) and identified as dodecanoic, 2,3-dimethylbutanoic, decanoic, hexanoic, tetradecanoic, and hexadecanoic acid.

Compound **1** was obtained as a pale yellow oil displaying a molecular ion [M]⁺ peak at *m/z* 854.6276 in HR-EI-MS, corresponding to the molecular formula C₅₂H₈₆O₉, suggesting 10 degrees of unsaturation in the molecule. In EI-MS, peaks at *m/z* 682 [M - 172]⁺, 738 [M - 116]⁺ and *m/z* 626 [M - 228]⁺ suggested it as a tri-ester of dodecanoic, 2,3-dimethylbutanoic, and tetradecanoic acid. The observed IR spectrum of **1** was assigned as 3534 (OH), 1767, 1754, 1745 (ester), 1704 (ketone), 1652 (C=C), and



Compound	R ¹	R ²	R ³	R ⁴
1	Bu	H	Tet	Dod
2	Dec	H	Hex	H
3	Bu	H	Hexd	Dod
4	H	H	Hex	Dod

Figure 1. Structures of compounds **1–4**. Bu, 2,3-di-Me-butanoyl; Dec, decanoyl; Dod, dodecanoyl; Hex, hexanoyl; Hexd, hexadecanoyl; Tet, tetradecanoyl.

1146 (C=O) cm⁻¹. The functional group moieties were further confirmed by ¹³C NMR (DEPT) spectral data (Table 2). The singlets appearing at δ 205.2 and 172–174 were attributed to the carbonyls of ketone and of ester, respectively. Four peaks observed in the middle region of the NMR spectrum at δ 128.4, 131.7, 136.3, and 137.2 were due to two C=C (tri-substituted) bonds in the skeleton. In the same spectrum, peaks in the region at δ 63–82 were due to the *gem*-hydroxy-acyloxy groups. In addition, four Me groups appeared at δ 22.6, 16.7, 15.5, and 18.3. The chemical shift, multiplicity, and coupling constant values in the ¹H NMR spectrum revealed that two Me groups are tertiary (δ 1.07 s, 1.21 s), one secondary (δ 0.96 d, *J* = 7.5 Hz), and one vinylic (δ 1.86 d, *J* = 1.2 Hz) in nature. In the HSQC spectrum, two Me singlets (δ 1.07, 1.21 s) showing correlations with carbons at δ 22.6, 16.7 were assigned to Me-16, 17, respectively; the third Me protons (δ 1.86) showed correlation with the carbon at δ 15.5 (Me-19), and the fourth Me protons (δ 0.96) with Me-18 resonated at δ 18.2. In the HMBC spectrum, Me-16, 17 showed correlations with C-15 at δ 30.5, indicating a *gem*-dimethyl moiety at C-15. Me-19 protons (δ 1.86) showed correlation with C-2 (δ 136.3), and Me-18 (δ 0.96) showed correlation with secondary C-11 resonated at δ 37.7.

The COSY-45° spectrum revealed two spin systems (i) CH⁷–CH⁸–CH¹⁴ and (ii) CH₂¹²–CH¹¹–CH₃¹⁸ in the molecule, indicating 13-hydroxyingenol as a parent system (Figure 2) [14–16].

NMR (¹H, ¹³C) spectra of **1** showed peaks for H-3 (δ 5.45), H₂-20 (δ 4.42, 4.73), C-3 (δ 80.2), and C-20 (δ 66.4), indicating ester moieties at these sites.

In the HMBC spectrum of **1**, H-3 showed correlation with the carbonyl carbon (δ 177.4) of 2,3-dimethylbutanoyl (Bu) (Figure 2) and the NOE experiment showed the interaction between H-3 and H-

Table 1. ^1H (300 MHz) and ^{13}C NMR spectral data of acid moieties for compounds **1** and **2** in CDCl_3 (δ , ppm, J in Hz).

Acid no.	Tetradecanoate		Dodecanoate		Decanoate		Hexanoate		2,3-Dimethylbutanoate	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1		174.3 s		174.3 s		173.3 s		172.3 s		176.3 s
2	2.30 t (6.2)	34.4 t	2.31 t (6.2)	34.4 t	2.32 t (6.2)	34.4 t	2.21 t (6.2)	36.3 t	2.19 m	46.2 d
3	1.25 br s	25.0 t	1.25 br s	25.0 t	1.25 br s	25.0 t	1.63 br s	16.3 t	1.83 m	30.0 d
4	1.24 br s	29.7 t	1.25 br s	29.7 t	1.24 br s	29.7 t	1.24 br s	16.2 t	0.85 d (7.2)	20.7 q
5	1.23 br s	29.7 t	1.25 br s	29.7 t	1.23 br s	29.7 t	1.23 br s	16.1 t	1.04 d (7.2)	13.9 q
6	1.23 br s	29.7 t	1.25 br s	29.7 t	1.23 br s	29.7 t	0.96 t (7.2)	13.7 t	0.84 d (7.3)	18.7 q
7	1.22 br s	29.7 t	1.25 br s	29.7 t	1.22 br s	29.7 t				
8	1.24 br s	29.7 t	1.25 br s	29.7 t	1.24 br s	29.7 t				
9	1.24 br s	29.7 t	1.25 br s	29.7 t	1.23 br s	22.7 t				
10	1.25 br s	29.7 t	1.25 br s	29.7 t	0.88 t (7.5)	14.2 q				
11	1.25 br s	29.7 t	1.23 br s	22.7 t						
12	1.25 br s	29.4 t								
13	1.23 br s	22.7 t								
14	0.88 t (7.2)	14.2 q								

Table 2. ^1H (300 MHz) and ^{13}C (75 MHz) NMR spectral data of compounds **1** and **2** in CDCl_3 (δ , ppm, J in Hz).

No.	1		2	
	δ_{H}	δ_{C}	δ_{H}	δ_{C}
1	6.01 q (1.2)	131.7 d	6.02 q (1.2)	131.8 d
2	—	136.3 s	—	136.5 s
3	5.43 s	82.8 d	3.87 s	80.7 d
4	—	74.7 s	—	74.6 s
5	3.83 br s	75.0 d	3.86 br s	74.9 d
6	—	137.2 s	—	136.9 s
7	5.63 d (4.6)	128.4 d	5.61 d (4.6)	128.2 d
8	4.14 dd (4.6, 12.5)	42.8 d	4.12 dd (4.6, 12.5)	42.6 d
9	—	205.2 s	—	205.5 s
10	—	72.0 s	—	72.1 s
11	2.45 ddq (4.5, 3.5, 7.5)	37.7 d	2.44 ddq (4.5, 3.5, 7.5)	37.8 d
12 α	2.83 ddd (15.5, 4.5, 3.5)	35.2 t	2.81 ddd (15.5, 4.2, 3.5)	35.1 t
12 β	2.37 dd (15.5, 4.5)	—	2.35 dd (15.5, 4.5)	—
13	—	69.2 s	—	69.1 s
14	1.64 d (12.4)	28.4 d	1.65 d (12.4)	28.5 d
15	—	30.4 s	—	30.3 s
16	1.21 s	22.6 q	1.23 s	22.5 q
17	1.07 s	16.7 q	1.07 s	16.6 q
18	0.96 d (7.2)	18.3 q	0.97 d (7.2)	18.1 q
19	1.86 d (1.2)	15.5 q	1.87 d (1.2)	15.4 q
20 α	4.421 d (12.5)	66.4 t	4.43 d (12.5)	66.2 t
20 β	4.72 d (12.5)	—	4.75 d (12.5)	—

2 (acyl part) concluding the attachment of 2,3-dimethylbutanoate at C-3. In the same spectrum, CH_2 -20 showed correlation with the carbonyl carbon of the dodecanoyl (Dod) moiety and the same was proved by

NOE. The third ester moiety, which did not display any interaction, was located at C-13, since the corresponding carbon (C-13) resonated in the low field at δ 69.2 in the ^{13}C NMR spectrum.

The ^{13}C NMR (BBD) spectrum of **1** showed 52 carbon atoms and DEPT spectra afforded 9 Me, 24 CH_2 and 9 CH, and 10 C. In the light of the spectral evidences, the structure of **1** was elucidated as 3-*O*-(2,3-dimethylbutanoyl)-13-*O*-tetradecanoyl-20-*O*-dodecanoyl-in-genol.

The relative stereochemistry of **1** was established with the help of the NOESY spectrum displayed in Figure 2.

Compound **2** was obtained as a colorless oil. IR and UV spectral data were identical to those of **1**. Compound **2** displayed a molecular ion peak [M^+] at m/z 616.3977 in HR-EI-MS corresponding to the molecular formula $\text{C}_{36}\text{H}_{56}\text{O}_8$. From the EI-MS fragmentation, peaks at m/z 516 [$\text{M} - 116$] $^+$,

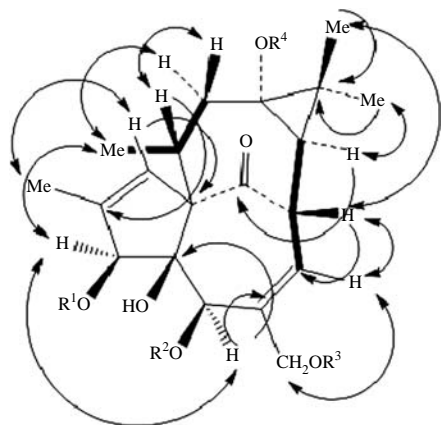


Figure 2. Key COSY and HOHAHA 45° (bold bonds), HMBC (curved arrows) and NOESY (dashed curved arrows) correlations of compounds **1**–**4**.

498, and 444 $[M - 172]^+$ indicated **2** as a diester of decanoic and hexanoic acid. In the NMR spectrum of **2**, H₂-20 resonated as an AB doublet at δ 4.43, 4.75 ($J = 12.5$ Hz), indicating the presence of one of the ester moiety at C-20. The HMBC spectrum showed the correlation of the carbonyl carbon of the hexanoyl moiety with H₂-20, suggested the attachment of the hexanoyl moiety at C-20, and the decanoyl carbonyl carbon displayed no interactions with any proton and hence placed at C-13. On the basis of the obtained results, **2** was named as 13-*O*-decanoyl-20-*O*-hexanoylingenol. The relative stereochemistry of all the stereogenic centers was established with the analysis of the NOE and NOESY spectra and found to be similar to that of **1** (Figure 2).

After comparing the available physical and spectral data of **3** and **4** with those in the literature [14,15], their structures were established as 3-*O*-(2,3-dimethylbutanoyl)-13-*O*-dodecanoyl-20-*O*-hexadecanoylingenol (**3**) and 13-*O*-dodecanoyl-20-*O*-hexanoylingenol (**4**). The relative stereochemistry of compounds **3** and **4** was also confirmed by recording the NOESY spectra and the measurement of NOE interaction. It was concluded that compounds **3** and **4** have the same stereochemistry as ingenols except for the presence of the ester moiety at C-13, having α -orientation.

In vitro cytotoxicity of isolates **1–4** was evaluated against RAW (mouse macrophage cells) and HT-29 (colon cancer cell lines) using the MTT [3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide] assay method. Compound **1** exhibited significant cytotoxicity with IC₅₀ 5.0 (RAW) and 2.90 (HT-29) μ M, concluding that **1** was 5- and 1.5-folds more potent against RAW and HT-29 cancer cell lines than standard amrubicin hydrochloride. Compound **2** showed moderate cytotoxic activity against both RAW and HT-29 cell lines (Table 3). The cytotoxicity of compound **4** against RAW

Table 3. *In vitro* cytotoxicity (IC₅₀, μ M) of compounds **1–4** against RAW and HT-29 cell lines.

Cell	Compounds				Standard
	1	2	3	4	
RAW	5.0	10.0	19.7	9.7	25.2
HT-29	2.9	10.0	15.4	15.4	4.4

cell lines was more potent than that of compound **2**. However, compound **3** showed no prominent activity against any cancer lines tested here.

3. Experimental

3.1 General experimental procedures

Optical rotation was measured on a digital polarimeter supplied by OSK OGAWA Seiki Co. Ltd, Tokyo, Japan. IR spectra were recorded in nujol mull ν_{\max} in cm^{-1} on a TENSOR 27 FT-IR spectrophotometer supplied by Bruker, Switzerland. ¹H and ¹³C NMR (300 and 75 MHz) spectra were obtained in CDCl₃ at room temperature, with TMS as an internal standard using a Bruker Biospin-AMX 300-MHz FT NMR spectrometer (Bruker); in δ , ppm, coupling constant J in Hz. Mass measurements were made on a double-focusing Finnegan MAT 112 spectrometer (Bremen, Germany) and recorded in m/z (relative intensity, %). HR-EI-MS measurements were obtained on a JEOL HX 110 spectrometer (JEOL, Tokyo, Japan).

3.2 Plant material

The *E. cornigera* Boiss. shoots were collected from Murree Hills, Pakistan, during the flowering season. The sample was identified by Prof. Dr Qazi Najm-u-Saqib, Ex-Dean Faculty of Pharmacy, GU, DIKhan and authenticated by comparing with samples present in the Herbarium of Botany Department, Peshawar University, Pakistan. A voucher specimen (DG/EP/06/113) has been deposited in the herbarium.

3.3 Extraction and isolation

The air-dried powdered shoots (3400 g) were extracted thrice with MeOH (7 liters) at room temperature, to give a brown residue (54.3 g). The material was suspended in MeOH:H₂O (6:1), and extracted successively with *n*-C₆H₁₄ (3 × 500 ml), CHCl₃ (3 × 500 ml), Et₂O (3 × 500 ml), EtOAc (3 × 500 ml), and *n*-BuOH (3 × 50 ml) to give corresponding 11.4, 5.3, 3.4, 2.3, and 1.2 g soluble materials, respectively. The Et₂O soluble material (3.3 g) was subjected to column chromatography over silica gel and eluted with the mixture of CH₂Cl₂:EtOAc:MeOH (9:1:0 → 0:1:9), to obtain nine crude fractions (A–I). All the fractions were tested against the cancer cell lines (RAW and HT-29), and only fractions **D** (137.2 mg) and **E** (134.3 mg) showed promising activities, and TLC results in CH₂Cl₂:EtOAc:MeOH (6:3:1) which displayed more than two components.

The active fractions were purified on preparative HPLC (Perkin-Elmer, Washington D.C., USA), HPLC RP-18 column (250 × 2.5 cm), after elution with a mixture of MeCN/H₂O, gradient scheme (A = H₂O; B = MeCN/H₂O 88/12; 5 min A, linear gradient to B in 20 min, 5 min B, linear gradient back to A in 2 min at 2 ml/min flow rate), as a mobile phase) to afford pure compounds **1** (23.3 mg), **2** (14 mg), **3** (14.4 mg), and **4** (79.1 mg) with retention times 7.3, 8.2, 9.2, and 10.7 min, respectively.

3.3.1 3-O-(2,3-Dimethylbutanoyl)-13-O-dodecanoyl-20-O-tetradecanoylingenol (**1**)

Pale yellow oil; $[\alpha]_D^{25} - 26.8$ ($c = 0.17$, CHCl₃). IR ν_{\max} (dry): 3534, 1767, 1754, 1745, 1724, 1652, 1146 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) and ¹³C NMR (75 MHz, CDCl₃) spectral data are listed in Table 1. EI-MS m/z : 854 [M, 3]⁺, 826 [M – 28, 13]⁺, 836 [M – 18, 13]⁺, 808 [M – 28 – 18, 23]⁺, 738 [M – 116, 12]⁺, 654 [M – 200, 28]⁺, 626

[M – 228, 22]⁺, 538 [M – 116 – 200, 23]⁺, 510 [M – 116 – 228, 32]⁺, 310 [M – 116 – 200 – 228, 72]⁺, 292 (40), 187 (57), 160 (75), 151 (82), 133 (53), 123 (54), 83 (100). HR-EI-MS m/z : 854.6276 [M]⁺ (calcd for C₅₂H₈₆O₉, 854.6272).

3.3.2 13-O-Decanoyl-20-O-hexanoylingenol (**2**)

Colorless oil, $[\alpha]_D^{25} - 33.7$ ($c = 0.17$, CHCl₃). IR (dry) ν_{\max} : 3503, 1754, 1735, 1625, 1552 cm⁻¹. ¹H NMR (300 MHz, CDCl₃) and ¹³C NMR (75 MHz, CDCl₃) spectral data are listed in Table 1. EI-MS m/z (rel. int.): 616 [M, 3]⁺, 598 [M – 18, 13]⁺, 588 [M – 28, 13]⁺, 570 [M – 18 – 28, 17]⁺, 500 [M – 116, 27]⁺, 482 [M – 116 – 18, 23]⁺, 444 [M – 172, 29]⁺, 426 [M – 172 – 18, 31]⁺, 312 (73), 294 (69), 284 (77), 251 (53), 221 (57), 188 (59), 162 (45), 153 (77), 151 (73), 135 (83), 122 (84), 121 (52), 83 (100). HR-EI-MS m/z : 616.3977 [M]⁺ (calcd for C₃₆H₅₆O₈, 616.3975).

3.4 Cytotoxicity assay

The cytotoxicity assay of samples **1–4** was evaluated on RAW and HT-29 cell lines, using the MTT assay as described previously [9,10]. The amrubicin hydrochloride salt was used as a control and all the measurements were taken in triplicate. Results in this connection are displayed in Table 3.

References

- [1] M. Shoeb, *Bangladesh J. Pharmacol.* **1**, 35 (Minireview), and references cited therein (2006).
- [2] I.B. Baloch, M.K. Baloch, and Q.N. Saqib, *Planta Med.* **72**, 830 (2006).
- [3] I.B. Baloch, M.K. Baloch, and Q.N. Saqib, *Eur. J. Med. Chem.* **43**, 274 (2008).
- [4] Y.B. Wang, R. Huang, H.B. Wang, H.Z. Jin, L.G. Lou, and G.W. Qin, *J. Nat. Prod.* **69**, 967 (2006).

- [5] L. Su, S.G. Feng, L. Qiao, Y.Z. Zhou, R.P. Yang, and Y.H. Pei, *J. Asian Nat. Prod. Res.* **11**, 38 (2009).
- [6] J.W. Yarbrow, The scientific basis of cancer chemotherapy, in *The Chemotherapy Source Book*, edited by M.C. Perry (Williams & Wilkins, Baltimore, MD, 1992), p. 2.
- [7] B.A. Chabner, Anti-cancer drugs, in *Cancer: Principles and Practice*, edited by V.T. DeVita Jr., S. Hellman, A.S. Rosenberg, 4th ed. (Lippincott, Philadelphia, MD, 1991), Chap. 18, p. 325.
- [8] J.M. Jessup, L.S. McGinnis, D.P. Winchester, H. Eyre, A. Fremgen, G.P. Murphy, and H.R. Menck, *CA Cancer J. Clin.* **46**, 185 (1996).
- [9] I.B. Baloch, M.K. Baloch, and Q.N. Saqib, *Helv. Chim. Acta* **88**, 3145 (2005).
- [10] I.B. Baloch, M.K. Baloch, Q.N. Saqib, and M. Ahmed, *Chem. Pharm. Med. J. Int.* **2**, 107 (2004).
- [11] I.B. Baloch, and M.K. Baloch, *J. Asian Nat. Prod. Res.* **12**, 600 (2010).
- [12] I.B. Baloch, M.K. Baloch, and A.K. Baloch, *Planta Med.* **76**, 809 (2010).
- [13] I.B. Baloch, M.K. Baloch, and A.K. Baloch, *Eur. J. Med. Chem.* **44**, 3188 (2009).
- [14] D. Uemura and Y. Hirata, *Tetrahedron Lett.* **29**, 2529 (1974).
- [15] T.T. Wu, Y.M. Lin, M. Haruna, D.J. Pan, and T. Shingu, *J. Nat. Prod.* **54**, 823 (1991).
- [16] Z.Q. Lu, M. Yang, J.Q. Zhang, G.T. Chen, H.L. Huang, S.H. Guan, C. Ma, X. Liu, and D.A. Guo, *Phytochemistry* **69**, 812 (2008).